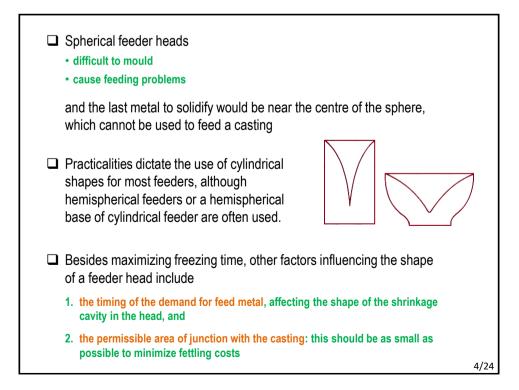
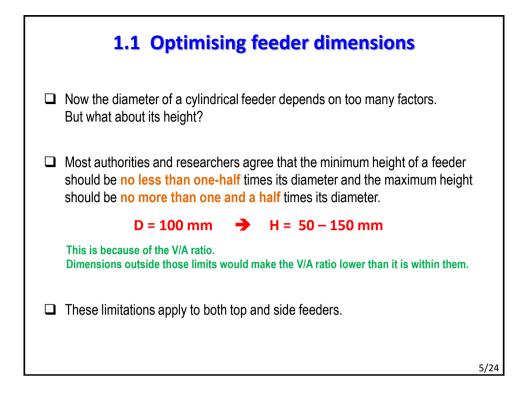
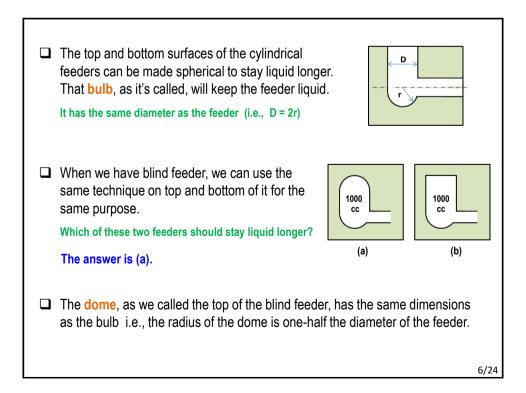


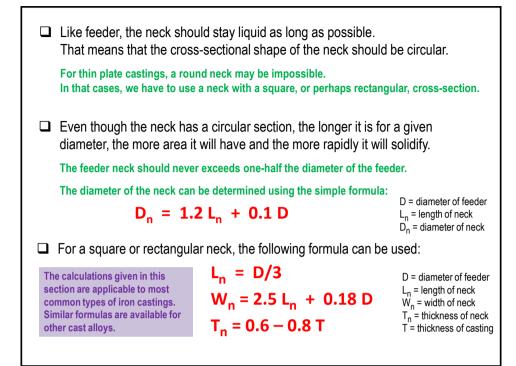
Amongst all shapes, a sphere has the smallest surface area thus a spherical feeder head will remain liquid for the longest period of time amongst all shapes of unit volume

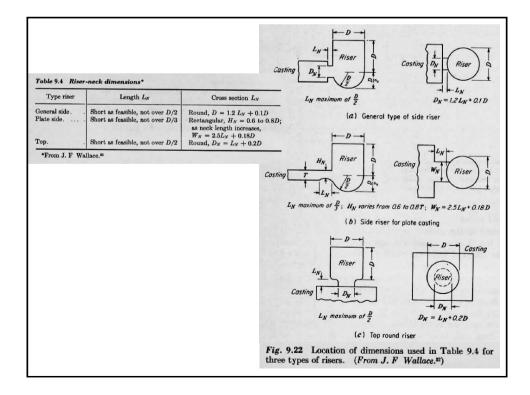
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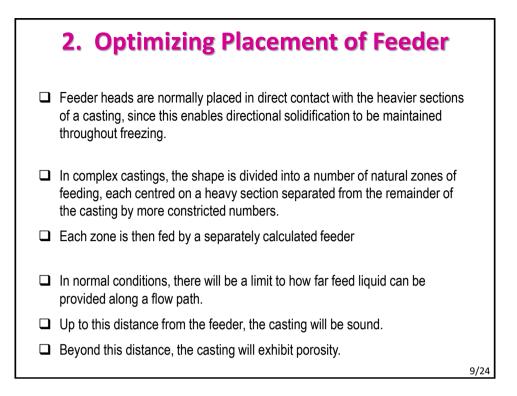


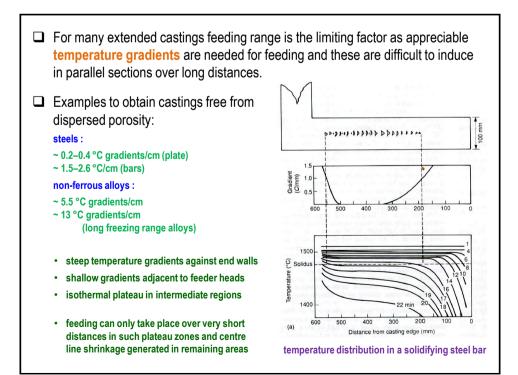


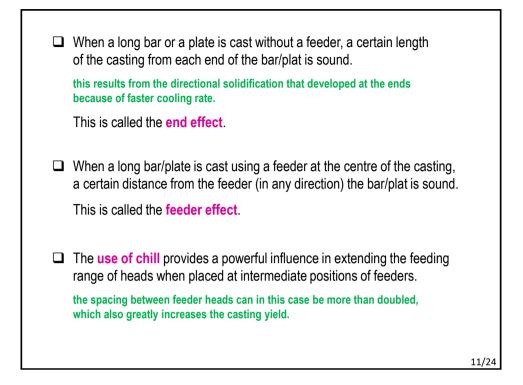


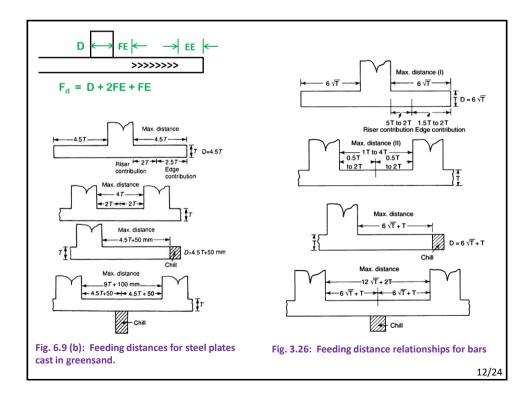


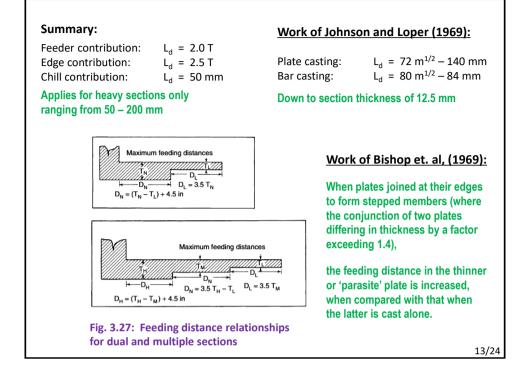


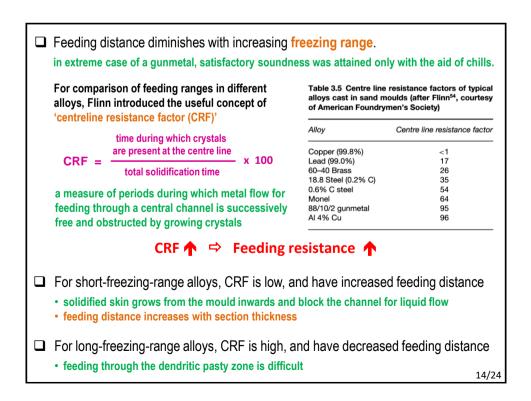


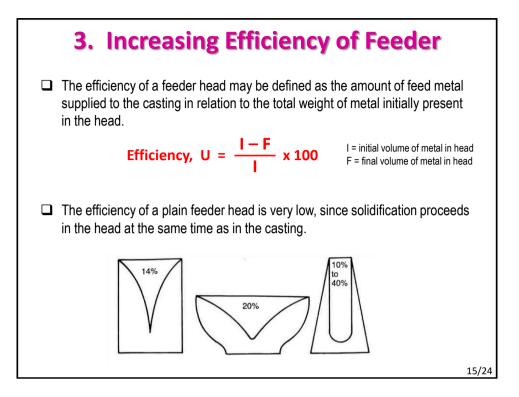


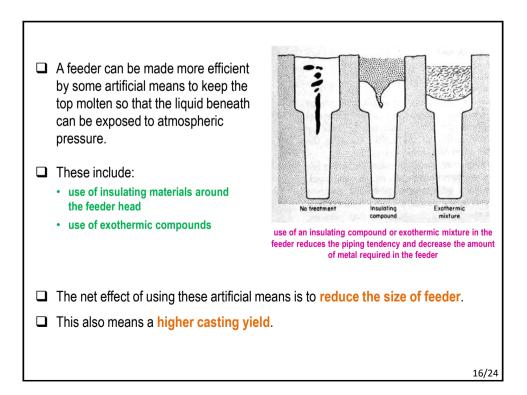


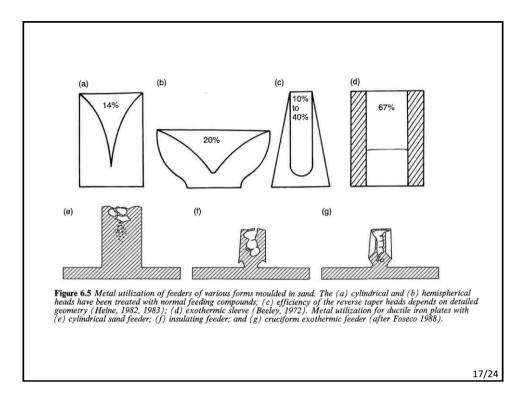












Insulation								
	The use of insulating materials to retard radiation losses from the exposed metal surface is a long established practice.							
	More thorough insulation is now frequently sought by lining the heads with moulded sleeves to reduce conductive heat loss through the mould.							
	The materials used derive their low heat diffusivity from porous or granular structures.							
 Top coverings materials include: dry sand, powdered slag, chopped straw (which first char and then burn away to leave a bulky ash), proprietary anti-piping compounds (contain a certain amount of exothermic material; those used of steel contains carbonaceous matter which becomes absorbed by the metal and locally reduces its freezing temperature) pre-formed sleeves for lining materials include: foamed gypsum plaster (to develop high porosity) (for non-ferrous castings), diatomaceous silica and vermiculite (for steel castings). 								
	Table 3.6 Feeder head insulation: freezing times for 4 in \times 4 in cylindrical feeder heads (min) (Reference 58)							
		No treatment	Top insulation only	Wall insulation only	Top and wall insulation			
	Steel Copp Alumi		13.4 14.0 14.3	7.5 15.1 31.1	43.0 45.0 45.6			
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Exothermic materials							
Considerable heat is generated by exothermic reaction; in some cases molten metal is also produced.							
Common materials used: thermit mixture (fine mixture of aluminium and iron oxide powdered charcoal or graphite, rice or oat husk, and refractory powder.),						
Thermit reactions:							
$2AI + Fe_2O_3 = AI_2O_3 + 2Fe;$ $\Delta H_{298} = -853 \text{ kJ}$ $8AI + 3Fe_3O_4 = 4AI_2O_3 + 9Fe;$ $\Delta H_{298} = -3347 \text{ kJ}$							
Added in two ways:							
1. Material is added on top of feeder to control feeding. Composition must be compatible with that of the casting.							
2. Material is mixed with bonding material and water to mould sleeve for lining feeder head. The added substance helps delaying exothermic reacting and extend the period during which heat is generated. This diminishes the danger of contamination, since the exothermic reaction is confined to the mould wall. The period during which heat is generated is controlled.							
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